

Regioselectivity in the Westphal Condensation

Adolfo Díaz, María P. Matia, José L. García-Navío,
Juan J. Vaquero, and Julio Alvarez-Builla*

Departamento de Química Orgánica, Universidad de Alcalá,
28871 Alcalá de Henares, Madrid, Spain

Received August 2, 1994

The Westphal condensation¹ is the basis of convenient easy methodology to build up quaternary aromatic nitrogen bridgehead systems by means of a condensation between α -methylcycloimmonium salts and 1,2-diketones.² This procedure has been applied to the synthesis of alkaloids such as sempervirine³ and flavocorylene⁴ which incorporate the zwitterionic indolo[2,3-*a*]quinolizinium ring system. In order to avoid the formation of regioisomers symmetrical 1,2-diketones have always been employed in the condensation.

Recently, we have been testing the method with unsymmetrical 1,2-diketones, and we present here the results obtained from the reaction of 1-aryl-1,2-propanediones (phenyl, 2-thienyl, *p*-tolyl, and 4-pyridyl) with some representative α -methylcycloimmonium substrates (Scheme 2).

Deprotonation of the starting salt **4** should produce the more stable N-ylide intermediate **5**, which should be the attacking nucleophile. In the presence of the diketone **7**, two alternative intermediates are possible. On kinetic grounds, the attack should take place at the more reactive methyl-substituted carbonyl group leading to structure **I** via the intermediate **8**. However, the thermodynamically controlled product **II**, passing through the more conjugated intermediate **9**, is the alternative possibility. The molar ratio **I**:**II** should be highly dependent on the electronic character of the aryl group in the starting diketone **7** ($R = Ar$). Thus, diketones bearing π -excessive arenes would increase conjugation in the intermediate **9** with the consequent rise in the proportion of **II**, while conversely, with π -deficient systems, kinetic control should favor the predominance of **I** in the final product.

The results in Table 1⁵ show how kinetic control predominates with 1-(4-pyridyl)- and 1-phenyl-1,2-propanediones, producing only 3-methyl derivatives **I**. In contrast, π -excessive aromatic systems such as 1-(2-thienyl)-, and to a lesser extent, 1-(*p*-tolyl)-1,2-propanedione, produced significant amounts of regioisomers **II**, as a consequence of the higher stability of the more conjugated intermediate. Moreover, when 2,3-pentanedione was used (entries 1, 6, 18, and 23) to demonstrate the

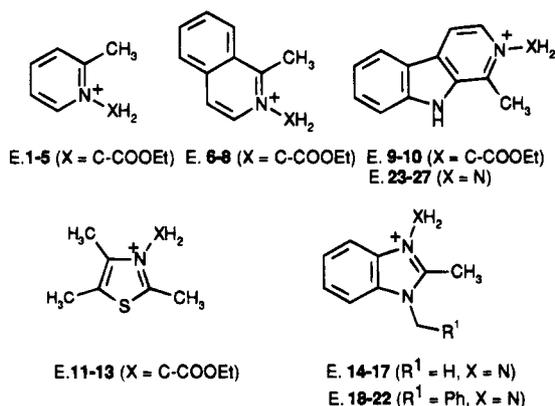
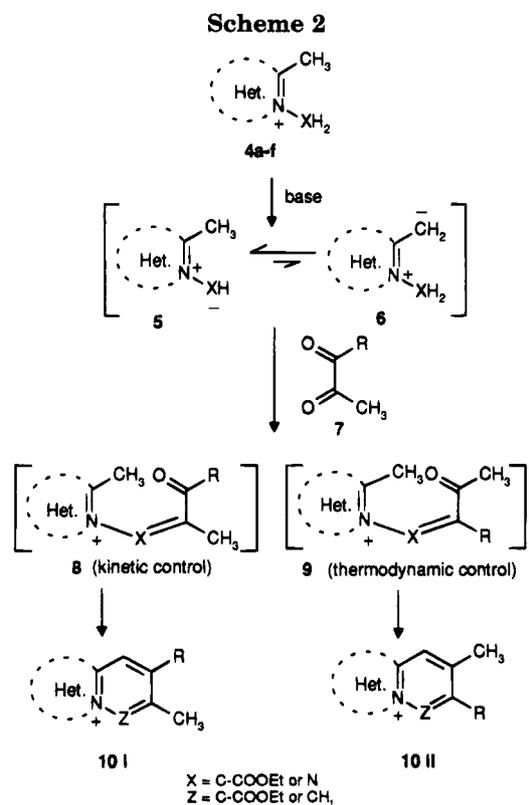
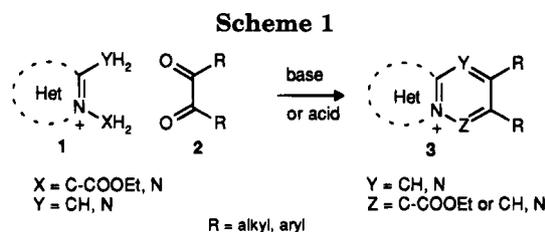


Figure 1. Substrates used in entries 1–22 in Table 1.



(1) Westphal, O.; Jahn, K.; Heffe, W. *Arch. Pharm.* **1961**, *294*, 37–45.

(2) (a) Baranova, N.; Sheinkman, A. K.; Kost, A. N. *Khim. Geterotsikl. Soedin.* **1973**, 1266–1270. (b) Matia, M. P.; Ezquerra, J.; Sanchez-Ferrando, F.; Garcia-Navio, J. L.; Vaquero, J. J.; Alvarez-Builla, J. *Tetrahedron* **1991**, *47*, 7329–7342. (c) G. Hajós, H.; Messmer, A.; Batori, S.; Riedl, Z. *Bull. Soc. Chim. Belg.* **1992**, *101*, 597–607. (d) Matia, M. P.; Garcia-Navio, J. L.; Vaquero, J. J.; Alvarez-Builla, J. J. *Liebigs Ann. Chem.* **1992**, 777–779. (e) Hajós, G.; Riedl, Z.; Gács-Baitz, E.; Messmer, A. *Tetrahedron* **1992**, *48*, 8459–8464.

(3) Potts, K. T.; Mattingly, G. S. *J. Org. Chem.* **1968**, *33*, 3985–3986.

(4) Matia, M. P.; Ezquerra, J.; Garcia-Navio, J. L.; Vaquero, J. J.; Alvarez-Builla, J. *Tetrahedron Lett.* **1991**, *32*, 7575–7578.

(5) Hydrolysis and decarboxylation of the ester group were observed in most of the examples, 1–13, as has been observed before; see ref 2b and references cited therein.

behavior of a diketone with no electronic effects differentiating between carbonyls, an approximately 50:50 mixture was formed.

The structures of the isolated products have been established using homonuclear NOE difference spectroscopy,⁶ employing a through-space connection between the protons, C1-H and C4-H, of the pyridinium, or C1-H of the pyridazinium moieties, and those of the methyl group directly attached to these fragments.

(6) Standard parameters were used, see: Holzer, W. *Tetrahedron* **1991**, *47*, 9783–9792.

Table 1. Westphal Condensations Performed To Obtain Compounds 10

entry	heterocycle	X	Z	R	react. time (h) solvent, temp	10 I:II ratio ^a	yield (%) ^b
1	2-pyridyl	CCOOEt	CH	ethyl	20, Me ₂ CO, rt	50:50 (11:12)	88
2	2-pyridyl	CCOOEt	CH	phenyl	15, Me ₂ CO, rt	100:0 (13)	67 ^c
3	2-pyridyl	CCOOEt	CH	<i>p</i> -tolyl	24, Me ₂ CO, rt	75:25 (14:15)	92
4	2-pyridyl	CCOOEt	CH	2-thienyl	24, Me ₂ CO, rt	40:60 (16:17)	88
5	2-pyridyl	CCOOEt	CH	4-pyridyl	2, Me ₂ CO, rfx	100:0 (18)	81 ^c
6	1-isoquinolyl	CCOOEt	CCOOEt	ethyl	3.5, Me ₂ CO, rfx	80:20 (19:20) ^d	75
7	1-isoquinolyl	CCOOEt	CH	phenyl	5, Me ₂ CO, rfx	100:0 (21)	45 ^c
8	1-isoquinolyl	CCOOEt	CH	2-thienyl	3.5, Me ₂ CO, rfx	50:50 (22:23)	85
9	1-(pyrido[3,4- <i>b</i>]indol)yl	CCOOEt	CH	phenyl	3, EtOH, rfx	100:0 (24)	70 ^e
10	1-(pyrido[3,4- <i>b</i>]indol)yl	CCOOEt	CH	4-pyridyl	2, EtOH, rfx	100:0 (25)	60 ^c
11	2-(4,5-dimethylthiazol)yl	CCOOEt	CH	phenyl	2.5, Me ₂ CO, rfx	100:0 (26)	57 ^c
12	2-(4,5-dimethylthiazol)yl	CCOOEt	CH	<i>p</i> -tolyl	2.5, Me ₂ CO, rfx	100:0 (27)	52 ^c
13	2-(4,5-dimethylthiazol)yl	CCOOEt	CH	4-pyridyl	2, EtOH, rfx	100:0 (28)	70 ^e
14	2-(3-methylbenzimidazol)yl	N	N	phenyl	2.5, EtOH, rfx	100:0 (29)	81 ^e
15	2-(3-methylbenzimidazol)yl	N	N	<i>p</i> -tolyl	2.5, EtOH, rfx	80:20 (30:31)	70
16	2-(3-methylbenzimidazol)yl	N	N	2-thienyl	3, EtOH, rfx	25:75 (32:33)	42
17	2-(3-methylbenzimidazol)yl	N	N	4-pyridyl	1, EtOH, rfx	100:0 (34)	76 ^e
18	2-(3-benzylbenzimidazol)yl	N	N	ethyl	3, EtOH, rfx	40:60 (35:36)	62
19	2-(3-benzylbenzimidazol)yl	N	N	phenyl	3.5, EtOH, rfx	100:0 (37)	60 ^e
20	2-(3-benzylbenzimidazol)yl	N	N	<i>p</i> -tolyl	5, EtOH, rfx	60:40 (38:39)	74
21	2-(3-benzylbenzimidazol)yl	N	N	2-thienyl	2, EtOH, rfx	33:66 (40:41)	88 ^e
22	2-(3-benzylbenzimidazol)yl	N	N	4-pyridyl	1, EtOH, rfx	100:0 (42)	79 ^e
23	1-(pyrido[3,4- <i>b</i>]indol)yl	N	N	ethyl	2.5, EtOH, rfx	50:50 (43:44)	86
24	1-(pyrido[3,4- <i>b</i>]indol)yl	N	N	phenyl	3, EtOH, rfx	100:0 (45)	37 ^c
25	1-(pyrido[3,4- <i>b</i>]indol)yl	N	N	<i>p</i> -tolyl	2, EtOH, rfx	66:33 (46:47)	85 ^c
26	1-(pyrido[3,4- <i>b</i>]indol)yl	N	N	2-thienyl	3, EtOH, rfx	25:75 (48:49)	91 ^c
27	1-(pyrido[3,4- <i>b</i>]indol)yl	N	N	4-pyridyl	2, EtOH, rfx	100:0 (50)	36 ^c

^aDetermined by ¹H NMR analysis of the crude reaction mixture. Compound identification is shown in brackets. ^bTotal yield in 10 I:II mixture. ^cExperimental procedure. Method A. ^dOverlapped signals. ^eExperimental procedure. Method B. Abbreviations: rt, room temperature; rfx, reflux.

In conclusion, the Westphal condensation, when performed using 1-aryl-1,2-propanediones, shows a typical kinetic vs thermodynamic control pattern, showing preference for the kinetically-controlled regioisomers. Mixtures of varying composition were obtained when π -excessive aryl groups were present in the 1,2-dicarbonyl fragment.

Experimental Section

Melting points are uncorrected. Mesitylenesulfonate anion signals are independent of the heterocyclic cations and appear at fixed values: 2.2 ppm for the *p*-methyl group, 2.4 ppm for the two *o*-methyl groups, and 6.7 ppm for the aromatic protons, and are not listed.

Starting heterocycles and 1,2-dicarbonyls were commercially available or obtained by previously described methods.⁷ The (ethoxycarbonyl)methyl salts **4a–d** (Scheme 2) were obtained as described for pyridine,^{8a} isoquinoline,⁹ β -carboline,⁴ and thiazole^{8b} derivatives, respectively. The *N*-amino salts **4e,g** were obtained as described for 1-methylbenzimidazole^{2d} and β -carboline³ derivatives. Only experiments allowing the isolation of single isomers are described.

1-Amino-3-benzyl-2-methylbenzimidazolium Mesitylenesulfonate (4f). To a stirred solution of amino mesitylenesulfonate (MSH) (0.97 g, 4.5 mmol) in CH₂Cl₂ (10 mL) was added 1-benzyl-2-methylbenzimidazole (1 g, 4.5 mmol) in CH₂Cl₂ (5 mL) dropwise. After 10 min of stirring at room temperature, Et₂O was added to precipitate the *N*-aminobenzimidazo-

lium salt, which was filtered and recrystallized from EtOH–Et₂O to give 1.48 g of **4f**, 75%: mp 195–198 °C; IR (KBr) 1654 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.89 (s, 3H); 5.77 (s, 2H); 6.65 (bs, 2H); 7.3–7.4 (m, 5H); 7.57 (t, 1H, *J* = 7.6 Hz); 7.63 (t, 1H, *J* = 7.6 Hz); 7.86 (d, 1H, *J* = 7.6 Hz); 7.90 (d, 1H, *J* = 7.6 Hz). Anal. Calcd for C₂₄H₂₇N₃O₃S: C, 65.87; H, 6.21; N, 9.60. Found: C, 65.52; H, 5.95; N, 9.28.

Westphal Condensation. General Procedure. Equimolar amounts of the corresponding azinium or azolium salt **4** (10 mmol), the dicarbonyl compound, and anhydrous sodium acetate were refluxed in the dry solvent indicated (10 mL), for the time described (Table 1). **Method A.** When a precipitate formed, the warm mixture was filtered and crystallization of the residue yielded the condensation product, which was analytically pure. **Method B.** Alternatively, when no precipitate was observed, the reaction mixture was concentrated to dryness, the residue triturated with acetone, and the mixture filtered. Recrystallization of the solid afforded the condensation product.

3-Methyl-2-phenylquinolizinium Bromide (13). Obtained by method A, 67%: mp 227–229 °C (from EtOH–Et₂O); IR (KBr) 1631 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.49 (s, 3H); 7.5–7.6 (m, 5H); 8.07 (t, 1H, *J* = 7.2 Hz); 8.33 (t, 1H, *J* = 7.2 Hz); 8.5–8.6 (m, 2H); 9.31 (d, 1H, *J* = 7.2 Hz); 9.50 (s, 1H). Anal. Calcd for C₁₆H₁₄BrN^{1/2}H₂O: C, 62.15; H, 4.89; N, 4.53. Found: C, 62.10; H, 4.92; N, 4.55.

3-Methyl-2-pyridylquinolizinium Bromide (18). Obtained by method A, 81%: mp 251–253 °C (from EtOH–Et₂O); IR (KBr) 1638 cm⁻¹; ¹H NMR (CD₃OD) δ 2.46 (s, 3H); 7.65 (d, 2H, *J* = 5.1 Hz); 8.12 (t, 1H, *J* = 6.6 Hz); 8.37 (t, 1H, *J* = 8.0 Hz); 8.53 (d, 1H, *J* = 8.4 Hz); 8.61 (s, 1H); 8.81 (d, 2H, *J* = 5.1 Hz); 9.32 (d, 1H, *J* = 6.3 Hz); 9.50 (s, 1H). Anal. Calcd for C₁₅H₁₃BrN^{1/2}H₂O: C, 56.44; H, 4.74; N, 8.78. Found: C, 56.25; H, 4.89; N, 8.61.

3-Methyl-2-phenylbenzo[*a*]quinolizinium Bromide (21). Obtained by method A, 45%: mp 337–339 °C (from EtOH–Et₂O); IR (KBr) 1639 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.54 (s, 3H); 7.6–7.7 (m, 5H); 8.00 (t, 1H, *J* = 8.3 Hz); 8.10 (t, 1H, *J* = 7.3 Hz); 8.25 (d, 1H, *J* = 8.0 Hz); 8.33 (d, 1H, *J* = 7.1 Hz); 8.91 (d, 1H, *J* = 7.3 Hz); 9.27 (d, 1H, *J* = 8.6 Hz); 9.33 (s, 1H); 9.55 (s, 1H). Anal. Calcd for C₂₀H₁₆NBr^{1/2}H₂O: C, 62.19; H, 5.22; N, 3.63. Found: C, 62.12; H, 5.51; N, 3.32.

3-Methyl-2-phenyl-12H-indolo[2,3-*a*]quinolizinium Bromide (24). Obtained by method B, 70%: mp >360 °C (from CH₃CO₂H); IR (KBr) 1633 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.51 (s,

(7) Both 2,3-pentanedione and 1-phenyl-1,2-propanedione are commercially available, 1-(2-thienyl)- and 1-(*p*-tolyl)-1,2-propanediones were prepared by nitrosation of the corresponding 1-substituted 1-propanones with isoamyl nitrite and final acid hydrolysis of the formed α -oximino ketones as described by Gianturco, M. A. et al. *Tetrahedron*, **1964**, 2951–2961. Finally, the 4-pyridyl derivative was obtained, as described by Knaus, E. E. *Can. J. Chem.*, **1980**, *58*, 130–133, starting with 4-pyridinecarboxaldehyde.

(8) (a) Alvarez-Builla, J.; Gonzalez Trigo, G.; Ezquerria, J.; Fombella, M. E. *J. Heterocycl. Chem.* **1985**, *22*, 681–685. (b) Galera, C.; Vaquero, J. J.; Garcia Navio, J. L.; Alvarez-Builla, J. *J. Heterocycl. Chem.* **1986**, *23*, 1889–1892.

(9) Ezquerria, J.; Alvarez-Builla, J. *J. Heterocycl. Chem.* **1986**, *23*, 1151–1157.

3H); 7.44 (t, 1H, $J = 7.8$ Hz); 7.6–7.7 (m, 6H); 7.80 (d, 1H, $J = 7.8$ Hz); 8.41 (d, 1H, $J = 7.8$ Hz); 8.78 (d, 1H, $J = 6.6$ Hz); 8.90 (s, 1H); 8.94 (d, 1H, $J = 6.6$ Hz); 9.43 (s, 1H); 13.35 (bs, 1H). Anal. Calcd for $C_{22}H_{17}N_2Br \cdot H_2O$: C, 64.87; H, 4.70; N, 6.88. Found: C, 64.52; H, 4.97; N, 6.56.

3-Methyl-2-pyridyl-12H-indolo[2,3-*a*]quinolizinium Bromide (25). Obtained by method A, 60%: mp >360 °C (from CH_3CO_2H); IR (KBr) 1633 cm^{-1} ; 1H NMR (DMSO- d_6) δ 2.53 (s, 3H); 7.48 (t, 1H, $J = 7.4$ Hz); 7.73 (t, 1H, $J = 7.4$ Hz); 7.86 (d, 1H, $J = 7.5$ Hz); 8.01 (d, 2H, $J = 4.8$ Hz); 8.46 (d, 1H, $J = 7.7$ Hz); 8.90 (d, 1H, $J = 7.5$ Hz); 9.0–9.1 (m, 4H); 9.53 (s, 1H); 13.48 (s, 1H). Anal. Calcd for $C_{21}H_{16}N_3Br \cdot \frac{1}{2}H_2O$: C, 63.17; H, 4.29; N, 10.52. Found: C, 62.93; H, 4.61; N, 10.73.

1,2,6-Trimethyl-5-phenylthiazolo[3,2-*a*]pyridinium Bromide (26). Obtained by method A, 57%: mp 273–275 °C (from EtOH–Et $_2$ O); IR (KBr) 1636 cm^{-1} ; 1H NMR (CD $_3$ OD) δ 2.53 (s, 3H); 2.69 (s, 3H); 2.72 (s, 3H); 7.5–7.6 (m, 5H); 8.46 (s, 1H); 9.04 (s, 1H). Anal. Calcd for $C_{16}H_{16}NSBr \cdot H_2O$: C, 54.55; H, 5.15; N, 3.97. Found: C, 54.53; H, 5.35; N, 3.69.

1,2,6-Trimethyl-5-*p*-tolylthiazolo[3,2-*a*]pyridinium Bromide (27). Obtained by method A, 52%: mp 251–253 °C (from EtOH–Et $_2$ O); IR (KBr) 1633 cm^{-1} ; 1H NMR (CD $_3$ OD) δ 2.44 (s, 3H); 2.54 (s, 3H); 2.69 (s, 3H); 2.71 (s, 3H); 7.39 (d, 1H, $J = 8.1$ Hz); 7.47 (d, 2H, $J = 8.1$ Hz); 8.44 (s, 1H); 9.02 (s, 1H). Anal. Calcd for $C_{17}H_{18}NSBr \cdot H_2O$: C, 58.62; H, 5.21; N, 4.02. Found: C, 58.37; H, 5.11; N, 4.42.

1,2,6-Trimethyl-5-pyridylthiazolo[3,2-*a*]pyridinium Bromide (28). Obtained by method B, 70%, mp 285–287 °C (from EtOH–Et $_2$ O); IR (KBr) 1627 cm^{-1} ; 1H NMR (CD $_3$ OD) δ 2.58 (s, 3H); 2.74 (s, 3H); 2.77 (s, 3H); 8.19 (d, 2H, $J = 6.6$ Hz); 8.71 (s, 1H); 9.03 (d, 2H, $J = 6.6$ Hz); 9.19 (s, 1H). Anal. Calcd for $C_{15}H_{15}N_2SBr \cdot H_2O$: C, 51.00; H, 4.85; N, 7.93. Found: C, 50.78; H, 5.00; N, 7.78.

2,5-Dimethyl-3-phenylpyridazino[1,6-*a*]benzimidazolium Mesitylenesulfonate (29). Obtained by method B, 81%: mp 235–238 °C (from EtOH–Et $_2$ O); IR (KBr) 1636 cm^{-1} ; 1H NMR (DMSO- d_6) δ 2.70 (s, 3H); 4.24 (s, 3H); 7.6–7.7 (m, 5H); 7.81 (t, 1H, $J = 7.7$ Hz); 7.94 (t, 1H, $J = 7.7$ Hz); 8.21 (d, 1H, $J = 8.3$ Hz); 8.44 (d, 1H, $J = 8.3$ Hz); 8.90 (s, 1H). Anal. Calcd for $C_{27}H_{27}O_3N_3S$: C, 68.47; H, 5.75; N, 8.87. Found: C, 68.38; H, 5.55; N, 8.76.

2,5-Dimethyl-3-pyridylpyridazino[1,6-*a*]benzimidazolium Bromide (34). After complexation of the refluxing time (Table 1), the solution was acidified with HBr drops (48%), and then the standard workup (method B) was carried out, 76%: mp 260–262 °C (from EtOH–Et $_2$ O); IR (KBr) 1636 cm^{-1} ; 1H NMR (DMSO- d_6) δ 2.73 (s, 3H); 4.29 (s, 3H); 7.8–7.9 (m, 3H); 8.00 (t, 1H, $J = 7.9$ Hz); 8.26 (d, 1H, $J = 8.4$ Hz); 8.50 (d, 1H, $J = 8.4$ Hz); 8.94 (d, 2H, $J = 5.1$ Hz); 9.09 (s, 1H). Anal. Calcd for $C_{17}H_{15}N_4Br \cdot H_2O$: C, 54.70; H, 4.59; N, 15.01. Found: C, 54.38; H, 4.76; N, 14.89.

5-Benzyl-2-methyl-3-phenylpyridazino[1,6-*a*]benzimidazolium Mesitylenesulfonate (37). Obtained by method B, 60%: mp 192–194 °C (from EtOH–Et $_2$ O); IR (KBr) 1626 cm^{-1} ; 1H NMR (DMSO- d_6) δ 2.57 (s, 3H); 6.10 (s, 2H); 7.3–7.5 (m, 5H); 7.6–7.7 (m, 5H); 7.81 (t, 1H, $J = 8.1$ Hz); 7.89 (t, 1H, $J = 8.3$ Hz); 8.06 (d, 1H, $J = 8.3$ Hz); 8.50 (d, 1H, $J = 8.1$ Hz); 9.13 (s, 1H). Anal. Calcd for $C_{33}H_{31}N_3O_3S \cdot H_2O$: C, 69.81; H, 5.86; N, 7.40. Found: C, 69.81; H, 5.54; N, 7.29.

5-Benzyl-3-methyl-2-(2-thienyl)pyridazino[1,2-*a*]benzimidazolium Mesitylenesulfonate (40). Obtained by method B, 60%: mp 218–220 °C (from CH_3CN); IR (KBr) 1636 cm^{-1} ; 1H NMR (DMSO- d_6) δ 2.88 (s, 3H); 6.07 (s, 2H); 7.3–7.4 (m, 4H);

7.5–7.6 (m, 2H); 7.79 (t, 1H, $J = 7.3$ Hz); 7.88 (t, 1H, $J = 7.6$ Hz); 7.9–8.0 (m, 2H); 8.09 (d, 1H, $J = 8.3$ Hz); 8.45 (d, 1H, $J = 8.0$ Hz); 9.26 (s, 1H). Anal. Calcd for $C_{31}H_{29}N_3O_3S_2 \cdot \frac{1}{2}H_2O$: C, 65.93; H, 5.35; N, 7.44. Found: C, 65.76; H, 5.18; N, 7.60.

5-Benzyl-2-methyl-3-(2-thienyl)pyridazino[1,2-*a*]benzimidazolium Bromide (41). This compound was isolated from the mother liquor of 39 by acidification of the acetone solution with HBr (48%), 28%: mp 285–287 °C (from EtOH); IR (KBr) 1600 cm^{-1} ; 1H NMR (DMSO- d_6) δ 2.98 (s, 3H); 6.12 (s, 2H); 7.3–7.5 (m, 6H); 7.79 (t, 1H, $J = 7.1$ Hz); 7.87 (t, 1H, $J = 7.1$ Hz); 7.96 (dd, 1H, $J = 3.8$ Hz, $J = 1.0$ Hz); 8.04 (d, 1H, $J = 8.3$ Hz); 8.09 (dd, 1H, $J = 2.6$ Hz, $J = 1.0$ Hz); 8.47 (d, 1H, $J = 8.1$ Hz); 9.16 (s, 1H). Anal. Calcd for $C_{22}H_{18}N_3SBr \cdot \frac{1}{2}H_2O$: C, 59.32; H, 4.30; N, 9.44. Found: C, 59.46; H, 4.17; N, 9.28.

5-Benzyl-2-methyl-3-pyridylpyridazino[1,6-*a*]benzimidazolium Mesitylenesulfonate (42). Obtained by method B, 79%: mp 185–187 °C (from EtOH–Et $_2$ O); IR (KBr) 1633 cm^{-1} ; 1H NMR (DMSO- d_6) δ 2.75 (s, 3H); 6.01 (s, 2H); 7.3–7.4 (m, 3H); 7.46 (d, 2H, $J = 8.5$ Hz); 7.78 (d, 2H, $J = 6.0$ Hz); 7.85 (t, 1H, $J = 7.5$ Hz); 7.92 (t, 1H, $J = 7.5$ Hz); 8.09 (d, 1H, $J = 8.5$ Hz); 8.53 (d, 1H, $J = 8.5$ Hz); 8.91 (d, 2H, $J = 6.0$ Hz); 9.24 (s, 1H). Anal. Calcd for $C_{32}H_{30}N_4O_3S$: C, 69.79; H, 5.49; N, 10.17. Found: C, 69.43; H, 5.68; N, 10.22.

3-Methyl-2-phenyl-12H-pyridazino[1',6':1,2]pyrido[3,4-*b*]indol-5-ium Mesitylenesulfonate (45). Obtained by method A, 37%: mp 294–296 °C (from CH_3CO_2H); IR (KBr) 1635 cm^{-1} ; 1H NMR (CF $_3$ CO $_2$ D) δ 2.75 (s, 3H); 7.4–7.5 (m, 6H); 7.6–7.7 (m, 2H); 8.24 (d, 1H, $J = 8.1$ Hz); 8.52 (d, 1H, $J = 6.8$ Hz); 8.78 (s, 1H); 8.98 (d, 1H, $J = 7.1$ Hz). Anal. Calcd for $C_{30}H_{27}N_3O_3S \cdot \frac{1}{2}H_2O$: C, 69.47; H, 5.44; N, 8.10. Found: C, 69.15; H, 5.19; N, 8.42.

3-Methyl-2-*p*-tolyl-12H-pyridazino[1',6':1,2]pyrido[3,4-*b*]indol-5-ium Mesitylenesulfonate (46). Obtained by method A, 46%: mp 307–309 °C (from CH_3CO_2H); IR (KBr) 1638 cm^{-1} ; 1H NMR (CD $_3$ OD) δ 2.49 (s, 3H); 2.83 (s, 3H); 7.47 (d, 2H, $J = 7.9$ Hz); 7.53 (t, 1H, $J = 8.1$ Hz); 7.62 (d, 1H, $J = 7.9$ Hz); 7.81 (d, 1H, $J = 9.0$ Hz); 7.83 (t, 1H, $J = 9.9$ Hz); 8.46 (d, 1H, $J = 8.0$ Hz); 8.84 (d, 1H, $J = 7.1$ Hz); 8.84 (d, 1H, $J = 7.1$ Hz); 8.98 (s, 1H); 9.18 (d, 1H, $J = 7.1$ Hz). Anal. Calcd for $C_{31}H_{29}N_3O_3S \cdot H_2O$: C, 68.74; H, 5.77; N, 7.76. Found: C, 68.46; H, 6.01; N, 7.42.

2-Methyl-3-(2-thienyl)-12H-pyridazino[1',6':1,2]pyrido[3,4-*b*]indol-5-ium Mesitylenesulfonate (49). Obtained by method A, 49%: mp >360 °C (from CH_3CO_2H); IR (KBr) 1638 cm^{-1} ; 1H NMR (DMSO- d_6) δ 2.83 (s, 3H); 7.16 (t, 1H, $J = 7.5$ Hz); 7.33 (t, 1H, $J = 4.4$ Hz); 7.47 (t, 1H, $J = 7.3$ Hz); 7.81 (d, 1H, $J = 8.5$ Hz); 7.9–8.0 (m, 2H); 8.32 (d, 1H, $J = 7.8$ Hz); 8.62 (d, 1H, $J = 6.8$ Hz); 8.73 (d, 1H, $J = 7.1$ Hz); 9.30 (s, 1H). Anal. Calcd for $C_{28}H_{25}N_3O_3S_2 \cdot H_2O$: C, 63.01; H, 5.10; N, 7.87. Found: C, 63.34; H, 4.99; N, 8.12.

3-Methyl-2-pyridyl-12H-pyridazino[1',6':1,2]pyrido[3,4-*b*]indol-5-ium Mesitylenesulfonate (50). Obtained by method A, 36%: mp 308–310 °C (from CH_3CO_2H); IR (KBr) 1622 cm^{-1} ; 1H NMR (DMSO- d_6) δ 2.79 (s, 3H); 7.54 (t, 1H, $J = 7.0$ Hz); 7.78 (d, 2H, $J = 5.5$ Hz); 7.81 (t, 1H, $J = 7.0$ Hz); 7.94 (d, 1H, $J = 7.9$ Hz); 8.60 (d, 1H, $J = 7.9$ Hz); 8.88 (d, 2H, $J = 5.5$ Hz); 9.07 (d, 1H, $J = 7.0$ Hz); 9.30 (s, 1H); 9.40 (d, 1H, $J = 7.0$ Hz). Anal. Calcd for $C_{29}H_{26}N_4O_3S$: C, 68.22; H, 5.13; N, 10.97. Found: C, 68.03; H, 5.42; N, 10.85.

Acknowledgment. The authors wish to thank the Comisión Interministerial de Ciencia y Tecnología (C.I.C.Y.T.) for financial support (Project PB90-0284).